

Tetramethylguanidinium chloride

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Key indicators

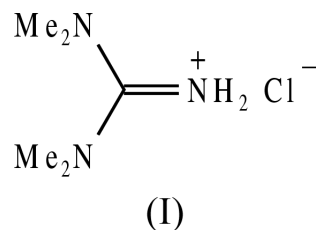
Single-crystal X-ray study
 $T = 143$ K
Mean $\sigma(\text{N}-\text{C}) = 0.002$ Å
 R factor = 0.025
 wR factor = 0.064
Data-to-parameter ratio = 20.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

In the title compound, $\text{C}_5\text{H}_{14}\text{N}_3^+\cdot\text{Cl}^-$, the central C atom displays almost ideal trigonal planar geometry. Classical hydrogen bonds of the form $\text{N}^+-\text{H}\cdots\text{Cl}^-$ link the formula units into discrete centrosymmetric dimers.

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Comment

The title compound was formed as an unexpected hydrolysis product during a study of guanidinophosphine derivatives (Münchenberg, 1996). Its structure is nevertheless of interest because few salts of this cation have been structurally investigated (CCDC refcodes: BEQQOD, KOYWEA, MATKAT and XERHEH; Version Oct. 2002; Allen & Kennard, 1993), and because of its hydrogen-bonding pattern.



The formula unit displays no imposed crystallographic symmetry. The geometry at the central atom C1 is almost ideal trigonal planar; the r.m.s. deviation from the least-squares plane of C1, N1, N2 and N3 is 0.006 Å. The methyl groups are rotated out of this plane (torsion angles are in Table 1).

The classical hydrogen bonds of the form $\text{N}^+-\text{H}\cdots\text{Cl}^-$ (Table 2) link the formula units into discrete centrosymmetric dimers (Fig. 1). Further contacts of the form $\text{C}-\text{H}\cdots\text{Cl}^-$, the shortest of which ($\text{C4}-\text{H4C}\cdots\text{Cl}$) has a normalized (Steiner, 1998) $\text{H}\cdots\text{Cl}$ distance of only 2.70 Å, may also be interpreted as hydrogen bonds.

Experimental

Crystal data

$\text{C}_5\text{H}_{14}\text{N}_3^+\cdot\text{Cl}^-$
 $M_r = 151.64$
Monoclinic, $P2_1/n$
 $a = 6.979$ (4) Å
 $b = 13.153$ (7) Å
 $c = 9.283$ (4) Å
 $\beta = 104.73$ (4)°
 $V = 824.1$ (7) Å³
 $Z = 4$

$D_x = 1.222$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 50 reflections
 $\theta = 10-11.5^\circ$
 $\mu = 0.39$ mm⁻¹
 $T = 143$ (2) K
Block, colourless
 $0.7 \times 0.6 \times 0.6$ mm

Data collection

Stoe Stadi-4 diffractometer
 ω/θ scans
 Absorption correction: none
 3151 measured reflections
 1900 independent reflections
 1763 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$

$\theta_{\text{max}} = 27.6^\circ$
 $h = -9 \rightarrow 9$
 $k = -17 \rightarrow 0$
 $l = -12 \rightarrow 6$
 3 standard reflections
 frequency: 60 min
 intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.025$
 $wR(F^2) = 0.065$
 $S = 1.08$
 1900 reflections
 94 parameters
 H atoms treated by a mixture of
 independent and constrained
 refinement

$w = 1/[\sigma^2(F_o^2) + (0.0265P)^2 + 0.1898P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.16 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.19 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

N1—C1	1.3304 (15)	N3—C1	1.3370 (14)
N2—C1	1.3417 (15)		
N1—C1—N3	120.91 (10)	N3—C1—N2	119.36 (10)
N1—C1—N2	119.70 (10)		
C4—N3—C1—N1	-23.93 (16)	C2—N2—C1—N1	-21.74 (16)
C5—N3—C1—N1	151.32 (11)	C3—N2—C1—N1	145.76 (11)

Table 2

Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N1—H1A \cdots Cl	0.918 (17)	2.415 (17)	3.2863 (17)	158.4 (13)
N1—H1B \cdots Cl ⁱ	0.866 (17)	2.373 (17)	3.2268 (17)	168.9 (14)
C3—H3B \cdots Cl ⁱⁱ	0.98	2.89	3.773 (2)	150
C4—H4A \cdots Cl ⁱⁱⁱ	0.98	3.00	3.776 (2)	137
C4—H4C \cdots Cl	0.98	2.77	3.574 (2)	139
C5—H5A \cdots Cl ^{iv}	0.98	2.96	3.922 (2)	167
C5—H5B \cdots Cl ⁱⁱⁱ	0.98	2.81	3.6492 (19)	144

Symmetry codes: (i) $1-x, 1-y, -z$; (ii) $\frac{1}{2}-x, \frac{1}{2}+y, \frac{1}{2}-z$; (iii) $x-1, y, z$; (iv) $1-x, 1-y, 1-z$.

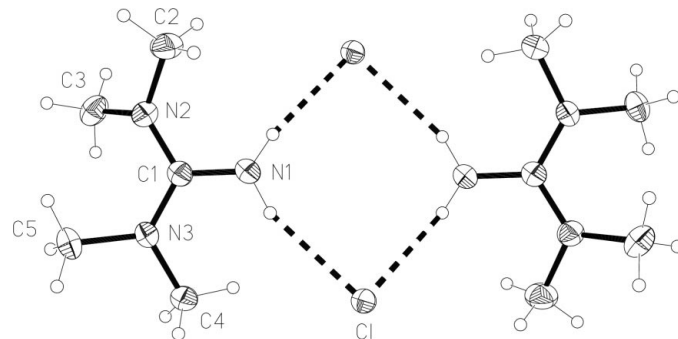


Figure 1

Dimeric unit of the title compound. Hydrogen bonds (Table 2) are indicated as thick dashed lines. Ellipsoids are at the 50% probability level.

H atom positions were obtained from difference syntheses. Methyl groups were refined as idealized rigid groups allowed to rotate about the C—N bonds. H atoms of the NH_2 group were refined freely.

Data collection: *DIF4* (Stoe & Cie, 1992); cell refinement: *DIF4*; data reduction: *REDU4* (Stoe & Cie, 1992); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* (Siemens, 1994); software used to prepare material for publication: *SHELXL97*.

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